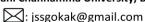
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Calendar of events prepared by IQAC



Calendar of Events 2022-23

The calendar of events was prepared as per the direction of Registrar, RCUB dated 15-07-2022

I and II-TERM

Sl. No.	Month	Activity
		➤ Start of I SEM admissions
		> Conduct of meeting on reopening day
		> Admission work (start of first sem NEP for the academic year 2022-23)
	1	> Preparation of time table and distribution of work
01	July-2022	> Induction programme for first sem students
	A Limited to a	Conduct of workshop on Candle Making, Preparation of Soaps and Detergents by Department of Chemistry for final year students
	, in the	Conduct of KSTA sponsored One Day National Conference on 'Recen Advances in Chemical Science' organized by Department of Chemistry
		> Commencement of I SEM classes
02	Aug-2022	➤ Celebration of Independence Day
	austras no s	➤ Observation of Librarian Day
		➤ Observation of Rhastriya Hindi Diwas
03	Sept-2022	➤ Observation of Teachers Day
	100000	➤ Observation of Gandhi Jayanthi
04	Oct-2022	➤ Commencement of III and V SEM classes
		Commencement of first internal test for I SEM *
		➤ Celebration of Karnataka Rajyothasava
05	Nov-2022	> Commencement of second internal test for I SEM
		Observation of Mahakavi Shri Kanakadas Jayanthi
		➤ End of I SEM classes
06	Dec-2022	➤ Commencement of I SEM Practical and Theory Examinations
		➤ Observation of National Mathematics Day





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07	Jan-2023	 ➤ Commencement of II SEM classes ➤ Observation of National Voters Day ➤ Celebration of Republic Day
08	Feb-2023	 End of III and V SEM classes Commencement of III and V SEM Practical and Theory Examinations Observation of National Science Day
09	Mar-2023	 ➤ Commencement of first internal test for II SEM ➤ Observation of International Women's Day ➤ Commencement of IV and VI SEM classes
10	Apr-2023	➤ Commencement of second internal test for II SEM ➤ Observation of Dr. B. R. Ambedkar Jayanthi
11	May-2023	 End of II SEM classes Commencement of first internal test for IV and VI SEM Sameelana: A gathering of Alumni and Parents organized jointly by Alumn Association and Parent-Teachers Cell
12	Jun-2023	 Commencement of second internal test for IV and VI SEM Observation of World Environmental Day NAAC Peer Team Visit
13	Jul-2023	Conduct of one-day workshop on 'Implications of IPR on startups' jointly organized by Department of Chemistry, Department of Commerce and Alumini Association
14	Aug-2023	 Conduct of one-day workshop on 'Research Methodology' jointly organized by Department Physics and Department of English Conduct of one-day workshop on 'Gender Equality' organized by Women Empowerment Cell Celebration of Independence Day Observation of Librarian Day Conduct of meeting to enter the IA marks in UUCMS portal Commencement of II, IV and VI SEM Practical and Theory Examination

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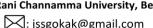
College Examination Committee (including Home assignment and IA Marks filling, sending to University) Signature Role Name of the Staff Sl. No. Convener Dr. U. M. Shagoti 01 Member Dr. Umesha B. 02 Member Prof. Abhay R. 03 Member Prof(Smt). D. A. Hulabandi 04 Member Prof. V. C. Hunachyali 05 Member Prof. A. M. Patil 06 Technical Staff Shri M. V. Dodamani 07 Technical Staff Smt. S. P.Jadhav (Tikkekar) 08 Smt. Vidyashree Shiragannavar Technical Staff 09 Smt Sunita. Kambar. 10

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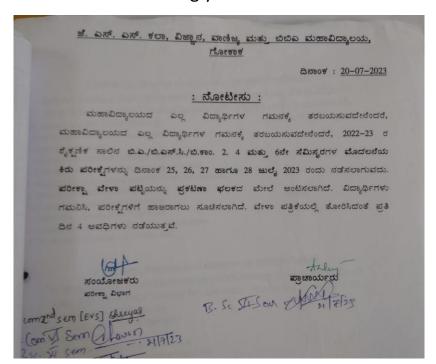
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Date: 20-07-2023

I.A. NOTICE

The II, IV and VI semesters students of B.A., B.Sc., and B.Com. are hereby informed that the first internal test will be commenced from 25, 26, 27 and 28 JULY, 2023. The time-table for the commencement of internal exam is displayed on the notice board. The students are asked to go through the time-table and attend accordingly.



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I.A. Time-table

Day/Date	Time		BA							
	0.00	II Sem	IV Sem	VI Sem	110	B.Com		B.Sc		
	9.00 to 10.00 am	Basic English	History- I	Economics-I	II Sem Basic English	IV Sem Hindi	VI Sem	II Sem Basic	IV Sem Botany	VI Sen Physics
Tuesday	11.30 to 12.30 pm	Basic Kan	Economics-I	Pol.Sc-I	******		Service	English Basic	Basic	Comp. So
25/7/2023	1.30 to 2.30 pm	History- I	Pol.Sc - I	Sociology P-I	Env.	Basic	Marketing Costing-II	Kannada Env.	English Comp. Sc	Zoology
	3.30 to 4.30 pm	Opt. Eng P-I	Opt. Eng P-I		Studies	Kannada		Studies	comp. sc	-
	9 to 10 am	Pol.Sc- I	IC	History- I	Fin. A/c-II	IC		Botany	Zoology	Chamists
Wednesday	11.30 to 12.30	Economics-I	Pol.Sc - II	Economics-II		Basic	Consumer	Chemistry	Chemistry	Chemistry Botany I
26/7/2023	1.30 to 2.30 pm	History- II	Economics-II	Opt.Kan P-I	Basic	English	Behaviour IF IM-	Maths	Maths	Comp. Sc
	3.30 to 4.30 pm	Opt. Kan P-I	Basic English	Opt. Hindi-I	Kannada 		*****	****		All SEC
	9 to 10 am	Sociology P-I	Sociology-I	Opt.Kan P-II	Law & Practice		Income Tax-	Comp. Sc	Basic	Subject
Thursday	11.30 to 12.30	Opt. Eng P-II	Opt. Kan P-I	Opt. Hindi-II Pol. Sc-II	of Banking Hindi / Urdu	Bus. Reg.	11	Physics	Kannada Physics	Chemistry Maths P
27/7/2023	1.30 to 2.30 pm	Pol. Sc-II	Opt. Eng P-II	Opt. Eng P-I		Frame Work	Auditing	Hindi /	Hindi /	Zoology F
	3.30 to 4.30 pm	Opt. Kan P-II	Basic Hindi/Urdu	History- II	7777	****		Urdu	Urdu	
	9 to 10 am	All OEC Subjects	Sociology-II	Opt. Eng P-II	Economics OEC - I & II	Costing Methods	*****	*****	*******	Botany F
Friday	11.30 to 12.30	Sociology P-II	History-II	All SEC Subjects	Corporate Administration	Corporate A/c	*****	Zoology	*****	Physics P
28/7/2023	1.30 to 2.30 pm	Economics-II	Basic Kan	Sociology P-II		A/C		All OEC	*****	Maths P
	3.30 to 4.30 pm	Basic Hindi/ Urdu	Opt. Kan P-II		2000		*****	Subjects		******

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I.A. Supervision list

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04 0	to P. M. Vadawad	4	4	4		4	4	1		4	V	V		4	V	V		
05 1	Mr. T. C. Gopal		*	4	4		4	4	4		V	*	*		4	V	*	
06 0	r K.I. Indikar	4	*		4	4	4		4	V	V		*	4	V			
07 3	shilpa T.	4	V		4	4	4		4	4	V		V	4	V			
08 p	r P. P. Kattimani		4	4	4		V	V	4		V	V	V		4	V	V	
000 5	hrt. H. S. Dasar	1	4		*	4	4		V	4	V		V	V	V		V	
10 8	hrt. M. F. Nagannavar		4	4	4		4	V	V		V	V	V		4	4	V	
11 8	hri. Kanakaraj T. M.		4	4	4		4	V	4		V	V	V		4	V	V	
12 8	hri Shiwaprasadagouda Patil		4	4	4		4	4	4		V	4	4		*	*	V	
13 0	r. N. L. Teradal	V	4	4	4	V	4	4	4	V	V	V	V					
14 0	on Rajestrusogi H.V.	V	4	4	V	V	4	V	4	4	4	4	4					
15 D	r. B. M. Mastiboli		4	4	4		4	4	4		4	4	4		V	V	V	
16 D	r. Vijayakumar Patil	4	4	4		4	4	*		V	*	1		4	4	4		
18 D	r. Krishnamurthy M. S.									4	*	4	4	4	4	4	¥	
19 D	r. Viswanath C. C.	4	4	4	4	4	4	4	4									
20 D	Chidanand Masaraguppi	4	*	4		4	*	4		1	4	1		4	4	4		
21 55	rs. Mahadevajipa Goragoddi	4	*	1		*	4	4		4	4	4		4	4	4		
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24 50	rt S. S. Chavan	4	*	4		V	4	*		4	*	V		4	4	4		
25 SN	rt A. K. Killikei	4	4	4		4	4	4		4	4	V		4	4	4		
6 Sh	rt. M. M. Katti		4	4	4		4	4	*		4	4	4		*	4	4	
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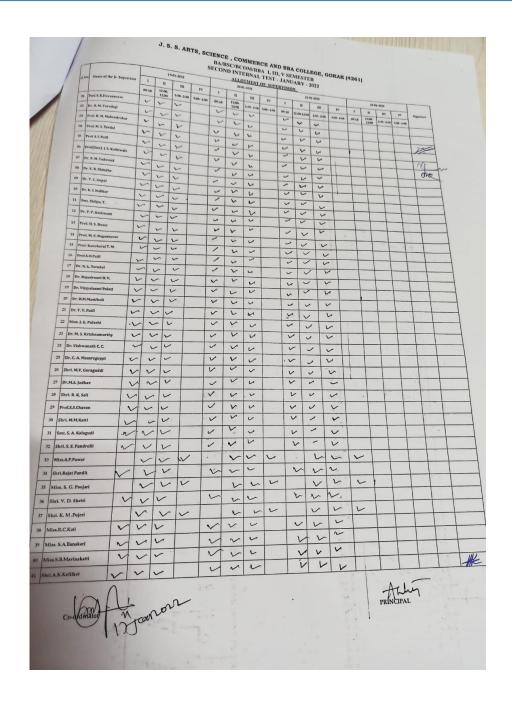
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PRACTICAL EXAMINATION FEBRUARY - 2022

TIME TABLE

NAME OF THE COLLEGE & CENTRE CODE: JSS ARTS, SCIENCE AND COMMERCE COLLEGE, GOKAK (4261)

CLASS:- B.Sc. V SEM

SUBJECT: PHYSICS

									Pape	er - II		
Date	Batch No.	Time : 9.30 to 1.30	No. of Students	Batch No.	Time : 2.00 to 6.00	No. of Students	Batch No.	Time: 9.30 to 1.30	No. of Student	Batch No.	Time : 2.00 to 6.00	No. of Student
10-02-2022	01	\$1917001, 005, 006, 007, 009, 010, 011, 012, 013, 014, 015,	12	02	\$1917 019, 020, 022, 023, 025, 029, 030, 031, 033, 035,	12	02	S1917019, 020, 022, 023, 025, 029, 030, 031, 033, 035, 036, 038	12	01	\$1917001, 005, 006, 007, 009, 010, 011, 012, 013, 014, 015, 016	12
11-02-2022	03	016 \$1917 040, 043, 044, 046, 047, 048, 049, 051, 053, 054, 056,	12	04	036, 038 \$1917059, 060, 062, 066, 067, 068, 070, 072, 073, 074, 079, 080	12	04	S1917059, 060, 062, 066, 067, 068, 070, 072, 073, 074, 079, 080	12	03	\$1917 040, 043, 044, 046, 047, 048, 049, 051, 053, 054, 056, 058	12
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16-02-20	022	17 S1917002, 069, 075, 084, 09 103, 114, 149, 167, 174, 176		18	\$1917185, 193, 221, 222	04	18	S1917185, 193, 221, 222	04		\$1917002, 069, 075, 084, 090, 103, 114, 149, 167, 174, 176, 184	12

Coordinator, Degree Practical Exam. February - 2022

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HOME ASSIGNMENT SAMPLE COPY

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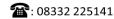
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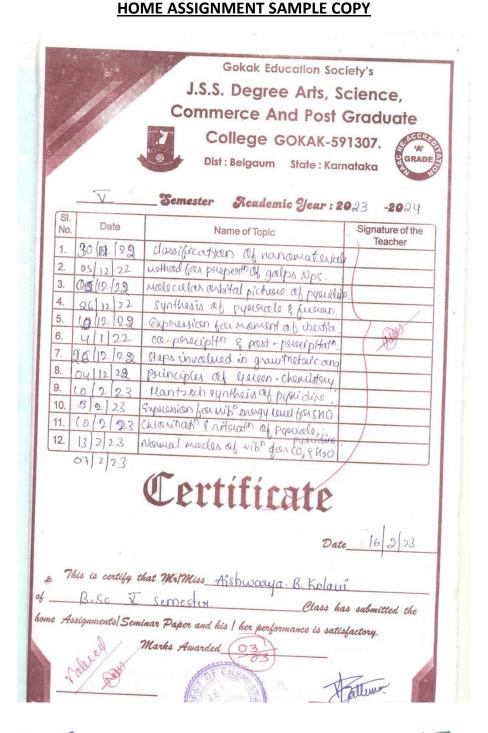
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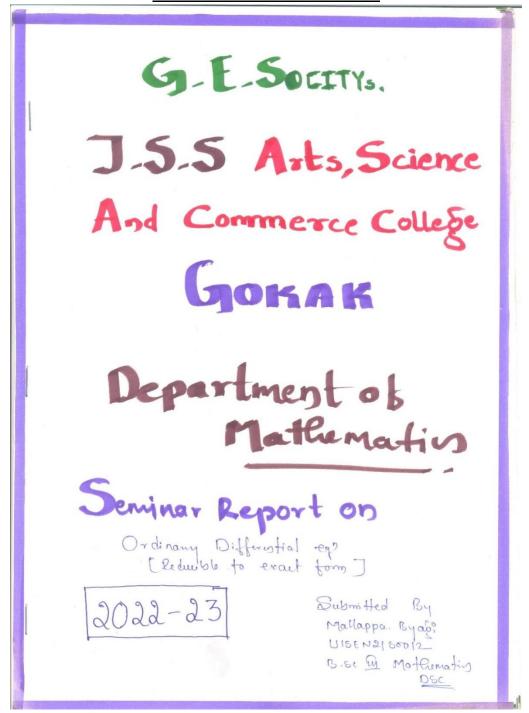
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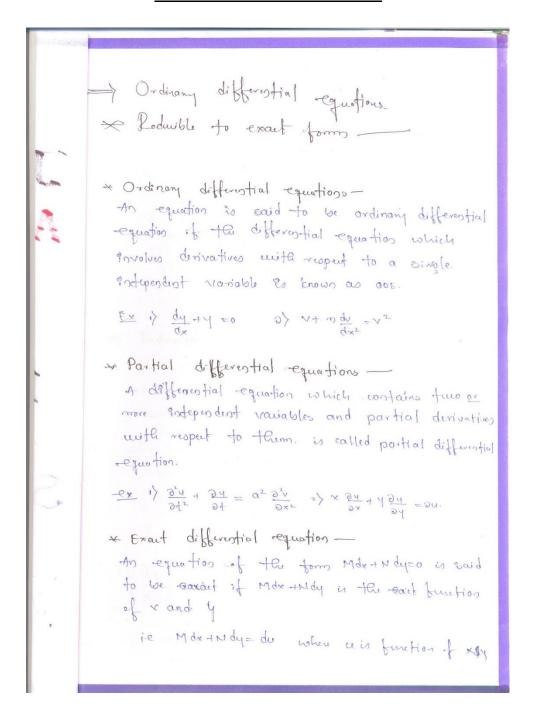
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G.E. Society's

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Department of Chemistry

> CERTIFICATE

This is to certify that the project work entitled "Preparation of Soaps and Detergents" is carried by Mr./Miss Arshuaya B Patil, , University Register Number 51917004 of B.Sc. VI Semester has completed and submitted the report on __Q\$ | q | 222__, as prescribed by Rani Channamma University, Belagavi during the year 2021-2022.

Staff Incharge

J.S.S. College, GOKAK A

Head of the Department CHEMISTRY J.S.S. College, Gokan

JOAC Co-ordinator
J.S.S. Arts, Science and Commerce College.



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SAMPLE COPY OF PROJECT WORK

ACKNOWLEDGEMENTS

It's indeed a great pleasure for me to express deep sense of gratitude to Dr. Rajeshweri H. V., Assistant Professor, Department of Chemistry, J. S. S. Arts, Science and Commerce College, Gokak, for giving hands-on training in preparing soaps and detergents and fruitful discussions throughout my project work.

I thank sincerely to *Dr. Smt. A. S. Terdal*, Principal, J. S. S. Arts, Science and Commerce College, Gokak and *Dr. R. S. Baligar*, Head of the Department of Chemistry, J. S. S. Arts, Science and Commerce College, Gokak for providing the required facilities to carry out my project work.

My sincere thanks to all teaching staff of Department of Chemistry, Dr. R. S. Baligar, Dr. P. P. Kattimani, Dr. N. L. Teradal, Dr. Krishnamurthy M. S., Dr. Viswanath C. C., Dr. Umesha B. Prof. C. B. Koujalagi, Smt. Anushree Mudholamath, for their continuous guidance, encouragement and timely help. I also thank non-teaching staff of the Department of Chemistry for their kind cooperation and support.

Finally, I take this opportunity to thank all those who have helped me directly or indirectly to complete this project work.

Thanking you,

Date: 08.19/22

Yours faithfully

(Mr/Miss/Mrs. Aishuaya B Patil)

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J.S.S. Arts, Science and Commerce College.
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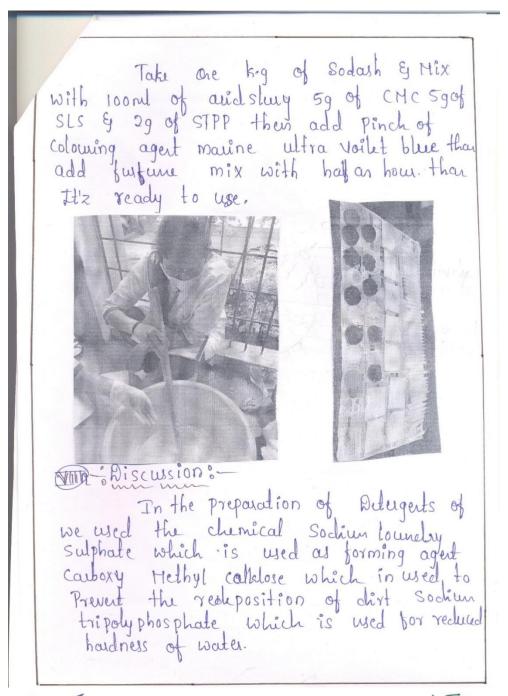
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SAMPLE COPY OF PROJECT WORK



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SAMPLE COPY OF INTERNAL ASSESSMENT QUESTION PAPER

J.S.S. Arts Science and Commerce College, Gokak

DEPARTMENT OF MATHEMATICS

I-IA Test July-2023

Class: II SEM SUB: Mathematics-(OEC) Marks = 30Section-A 1) Answer any five of the following 2X5=10 1) Define Group 2) Define Abelian Group 3) Define Subgroup 4) Define Cyclic Group 5) State Euler's Extension theorem for homogeneous functions 6) If $u = (x^2 + y^2 + z^2)^{\frac{-1}{2}}$ then find $\frac{\partial^2 u}{\partial x^2}$ Aprel trions tove A Answer any five of the following 4X5 = 207) Show that Identity element in a group is unique 8) Show that the inverse of each element of a group is unique 9) State and Prove Euler's Theorem 10) If $u = \log(x^2 + y^2)$ then show that $x \frac{\partial u}{\partial x} + y \frac{\partial u}{\partial y} = 2$ 11) If $u = \sin^{-1}\left(\frac{x^2+y^2}{x+y}\right)$ then show that $x\frac{\partial u}{\partial x} + y\frac{\partial u}{\partial y} = \tan u$ 12) If u=xy with x=at² and y=2at then find $\frac{du}{dt}$

J.S.S. Arts Science and Commerce College, Gokak DEPARTMENT OF MATHEMATICS

I -IATest July-2023

Class:VI SEM

SUB: Mathematics-I

Marks = 20

- Answer any four of the following questions
- 5 X 4=20

- 1) a) Define harmonic function.
 - b) If f(z) = u + iv be an analytic in domain D then u & v satisfies C-R equation then show that $u_x = v_y \& u_y = -v_x$
- 2) a) Prove that analytic function with constant imaginary part is constant
 - b) Show that $\frac{\partial(u,v)}{\partial(x,y)} = |f'(z)|^2$ if f(z) = u + iv is constant
- 3) Prove that an analytic function with constant modulus is constant
- 4) If $u(x,y) = x^3 3xy^2$ then show that it is harmonic find its harmonic find its harmonic conjugate and corresponding analytic function
- 5) State and prove Cauchy's integral theorem or Fundamental Cauchy's theorem
- 6) State and prove Cauchy integral formula for derivatives

IQAC Co-ordinator

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J.S.S. Arts, Science and Commerce College, Gokak

NAAC accredited at 'A' grade with CGPA 3.15 in 4th Cycle

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SAMPLE COPY OF INTERNAL ASSESSMENT QUESTION PAPER

J.S.S. ARTS, SCIENCE AND COMMERCE COLLEGE, GOKAK

II-IA 2022-23 Subject: Chemistry (NEP) B.Sc. II Semester (DSC)

Time: 1 hour

Maximum Marks: 30 Marks

Answer Any Seven of the following:
 Write the structure of ammonia molecule and mention the bond angle

Write the difference between bonding and antibonding molecular orbitals What is Arrhenius acid and base? Give example for each.

What is Bronsted acid and base? Give example for each.Write the difference between crystalline and amorphous solid.

What are liquid crystals? Give example. What is unit cell and space lattice?

Define law of rational indices.

II. Answer the following questions:

4 X 4 = 16

a) Discuss the shape of H₂O molecule on the basis of VSEPR theory.

OR b) Write the salient features of MOT.

10. a) Explain factors affecting pKa values of organic acids.

OR

(b) Explain relative acidic strength of the followings; Chloroacetic acid, acetic acid and propanoic acid

11. a) Derive Bragg's equation.

OR

b) Explain the types of unit cell and calculate the number of particles per unit cell.

12. a) Explain the law of constancy of symmetry. OR

b) Explain the properties of liquid crystals.

J. S. S. Arts, Science and Commerce College, Gokak

II-IA 2022-23

Subject: Chemistry (DSC) Time: 1 hour

B. Sc. V Semester (Paper-I) Maximum Marks: 30 Marks

2x7 = 14 Marks

I. Answer any Seven of the following

What are ligands? Give examples.
 Write the coordination number of central metal ion in i) [Cu(en)₂]SO₄ ii) [CoCl₂(en)₂]Cl
 What is a children in the Circumstance of t

 What is polydentate ligand? Give an example
 Write the criteria to exhibit vibrational spectroscopy? 5. Write the selection rule for vibrational spectroscopy.
5. Write the selection rule for vibrational spectroscopy.
6. What is zero-point energy?
7. What are polymers? Give examples.
8. What are thermosetting polymers? Give an example.

II. Answer the following

I. Answer the following

9. (a) Mention the type of hybridization for: (a) Linear (b) Square planar (c) Tetrahedral (d) Octahedral geometry

OR
(b) Write the IUPAC names for: a). K₄[Fe(CN)₆] b) [Zn(NH₃)₄]SO₄
c) [Ni(CN)₄]² d) [Co(NH₃)₃Cl₃]

10. (a) What are inner orbital complexes and outer orbital complexes. OR
(b) Derive an expression for the determination of force constant in vibrational spectroscopy.

11. (a) Derive an energy level expression for simple harmonic oscillator and write vibrational energy level diagram. OR level diagram. OR

(b) Sketch and calculate the vibrational degrees of freedom for CO₂ and H₂O molecule.

12. (a) Explain the classification of polymers OR

(b) Explain the mechanical properties of polymers.

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IQAC Co-ordinator J.S.S. Arts, Science and Commerce College. Gokal.